

Krane Chapter 6

Problem 8

$$d = (e^2/4\pi\epsilon_0)Zz/K = (1.44\text{Mev} - \text{fm})(2)(29)/6.0\text{Mev} = 14\text{fm}$$

Problem 17

The transitions of the Lyman series are those with $n_0 = 1$ Paschen series have $n_0 = 3$

$$\Delta E = E_n - E_{n_0} = h\nu = hc/\lambda$$

$$\lambda = (hc/13.6\text{eV})(n \cdot n_0)^2/(n^2 - n_0^2)$$

$$\lambda_{\text{limit}} = (hc/13.6\text{eV})(n_0)^2 \quad (\text{Let } n \text{ goto infinity})$$

Problem 18)

$$\text{From } v = n\hbar/mr = n\hbar/(mn^2a_0) = (1/\hbar n)((e^2/4\pi\epsilon_0) = \alpha c/n$$

For a different nuclear charge use Ze

$$v = Z\alpha c/n$$

Problem 22

The photon energies of the incident light are

$$E = hc/\lambda = 1240\text{eV} - nm/59nm = 21\text{eV}$$

When (and if) an atom in a ground state absorbs a 21 eV photon the atom is ionized (requires only 13.6 eV).

The excess energy (21-13.6) appears as kinetic energy of the electron (neglecting the small recoil energy of the nucleus.) Hence 21-13.6 = 7.4 eV is the KE of the electron.

Problem 37

$$(a) \quad 2\pi r = n\lambda, \quad n = 1, 2, 3, \dots$$

If the circumference is an integral number of deBroglie waves, then after each orbit there will be a "wave with no beats" or a standing wave on the circle.

$$(b) \text{ with } \lambda = h/p$$

$$2\pi r = n\lambda = nh/p = nh/mv$$

so

$$mvr = nh/2\pi = n\hbar$$